MATERIAL SAFETY DATA SHEET

1. SUBSTANCE AND SOURCE IDENTIFICATION

National Institute of Standards and Technology

Standard Reference Materials Program 100 Bureau Drive, Stop 2320

Gaithersburg, Maryland 20899-2320

SRM Number: 4325 MSDS Number: 4325

SRM Name: Beryllium-10/Beryllium-9

Isotopic Ratio Standard

Date of Issue: 02 May 2005

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Description: SRM 4325 contains Beryllium-10/Beryllium-9 chloride in 50-milliliters of

The solution is packaged in two 30-milliliter TeflonTM bottles (25 mL each). The solution consists of 5.15 mg of beryllium per gram of 1 N hydrochloric acid. The SRM is intended for the calibration of accelerator

mass spectrometers used to measure beryllium isotopic ratios.

Substance: Beryllium-10/Beryllium-9 Chloride in 1 N Hydrochloric Acid

Other Designations: Beryllium-10/Beryllium-9 Chloride (beryllium dichloride; glucinium chloride)

in 1 N Hydrochloric Acid (aqueous hydrochloric acid; hydrogen chloride;

muriatic acid).

2. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

Component: Hydrochloric Acid

CAS Number: 7647-01-0 **EC Number (EINECS):** 231-595-7

SRM Nominal

Concentration: 1 N T, C **EC Classification:** EC Risk (R): 23, 35

> EC Safety (S): 1, 2, 9, 26, 36, 37, 39, 45

Beryllium Chloride (powder form) Component:

CAS Number: 7787-47-5 **EC Number (EINECS):** 232-116-4

SRM Nominal

Concentration (mass %): 0.515 **EC Classification:** T, Xi, N

> EC Risk (R): 23, 25, 26, 36, 37, 38, 43, 48, 49, 51, 53

EC Safety (S): 45, 53, 61

3. HAZARDS IDENTIFICATION

Hydrochloric Acid

NFPA Ratings (Scale 0-4): Health = 3Fire = 0Reactivity = 0

Beryllium Chloride

NFPA Ratings (Scale 0-4): Health = 3Fire = 0Reactivity = 2

MSDS 4325 Page 1 of 6 **Potential Health Effects**

Inhalation: Inhalation of hydrochloric acid fumes may cause irritation and burning of the

nose, throat, and upper respiratory tract, coughing and choking.

Skin Contact: Skin contact of hydrochloric acid and beryllium chloride may cause severe

irritation. Skin contact of hydrochloric acid may also cause inflammation and

chemical burns.

Eye Contact: Hydrochloric acid vapors are irritating and may cause damage to the eyes. Eye

contact of hydrochloric acid and beryllium chloride may cause severe irritation,

conjunctivitis, burning sensation, and possible corneal damage.

Ingestion: Ingestion of hydrochloric acid can cause pain and burns of the mouth, throat,

esophagus, and stomach. May also cause nausea, vomiting, diarrhea, chills, shock, and intense thirst. Perforation of the intestinal tract and circulatory collapse may occur. Death may occur due to esophageal or gastric necrosis. Animal studies of acute ingestion of beryllium chloride indicate that low doses

may be fatal.

Major Health Hazards: Hydrochloric Acid: Respiratory tract burns. Skin burns. Eye burns. Mucous

membrane burns.

Listed as a Carcinogen/ Possible Carcinogen Hydrochloric Acid:

Yes No

X
 X
 In the National Toxicology Program (NTP) Report on Carcinogens.
 In the International Agency for Research on Cancer (IARC) Monographs.

By the Occupational Safety and Health Administration (OSHA).

Beryllium Chloride: Yes No

X In the National Toxicology Program (NTP) Report on Carcinogens.

In the International Agency for Research on Cancer (IARC) Monographs.

By the Occupational Safety and Health Administration (OSHA).

4. FIRST AID MEASURES

Skin Contact: Rinse affected area with copious amounts of water for at least 15 minutes while

removing contaminated clothing followed by washing the area with soap and

water. Obtain medical assistance if necessary.

Eye Contact: Immediately flush eyes, including under the eyelids, with copious amounts of

water for at least 15 minutes. Obtain medical assistance immediately.

Inhalation: If adverse effects occur, remove to uncontaminated area. Give artificial

respiration by qualified personnel if not breathing. Get medical attention if

necessary.

Ingestion: If ingestion occurs, immediately contact poison control center or physician. **DO**

NOT INDUCE VOMITING. Give large quantities of water. Never give anything by mouth to an unconscious person. If vomiting occurs, keep head lower than hips to prevent aspiration. If a person is unconscious, turn head to

side. Obtain immediate medical assistance.

5. FIRE FIGHTING MEASURES

Fire and Explosion Hazards: Hydrochloric acid and beryllium chloride are negligible fire hazards.

Extinguishing Media: Use regular dry chemical, carbon dioxide, water, or regular foam.

Fire Fighting: DO NOT touch spilled material. Move container from fire area if it can be done without risk. Use extinguishing agents appropriate for surrounding fire. Avoid

inhalation of material or combustion by-products.

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Flash Point (°C): Not applicable.

Method Used: Not applicable.

Autoignition Temp. (°C): Not applicable.

Flammability Limits in Air

UPPER (Volume %): Not applicable. **LOWER (Volume %):** Not applicable.

6. ACCIDENTAL RELEASE MEASURES

Occupational Release: DO NOT touch spilled material. Notify safety personnel of spill. Absorb with

sand or other non-combustible material. Collect with absorbent into suitable

container for proper disposal

Disposal: Refer to Section 13, "Disposal Considerations".

7. HANDLING AND STORAGE

Storage: Store and handle in accordance with all current regulations and standards. Keep

separated from incompatible substances. Store in a well-ventilated area.

Safe Handling Precautions: See Section 8, "Exposure Controls and Personal Protection".

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Exposure Limits: Hydrochloric Acid

OSHA (PEL): 7 mg/m³ (5 ppm) ceiling

ACGIH (TLV): 2 ppm ceiling

NIOSH: 7 mg/m³ (5 ppm) recommended ceiling OES UK: 2 mg/m³ (1 ppm) TWA (gas, mist) OES UK: 8 mg/m³ (5 ppm) STEL (gas, mist)

Beryllium Chloride

OSHA (PEL): 2 μg (Be)/m³ TWA OSHA (PEL): 5 μg (Be)/m³ ceiling ACGIH (TLV): 0.002 mg (Be)/m³ TWA ACGIH (TLV): 0.01 mg (Be)/m³ STEL

NIOSH: 0.0005 mg (Be)/m³ (10 h) recommended TWA

UK MEL: 0.002 mg (Be)/m³ TWA

Ventilation: Use a local exhaust ventilation system. Ensure compliance with applicable

exposure limits.

Eve Protection: Wear safety goggles or a face shield. **DO NOT** wear contact lenses in the

laboratory. An eye wash station should be readily available near areas of use.

Personal Protection: Wear appropriate protective clothing and disposable chemically resistant gloves

to prevent skin exposure.

9. PHYSICAL AND CHEMICAL PROPERTIES

Component: Hydrochloric Acid

Appearance and Odor: Colorless, liquid. Pungent odor.

Relative Molecular Weight: 36.46 g/mol

Molecular Formula: HCl

Evaporation Rate

(HCl solutions 0.02 N to 2 N):> 1 (ether = 1)

Density (1 N HCl): 1.0 g/cm³ Soluble.

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Component: Beryllium Chloride (powder form)

Appearance: Colorless to yellow solid.

Relative Molecular Weight: 79.91 g/mol **Molecular Formula:** BeCl₂

Density: 1.899 g/cm³
Water Solubility: Soluble.

10.	STABILITY	AND REACTIVITY
IV.	SIADILIII	AND INDACTIVITY

Stability: X Stable Unstable

Stable under ordinary conditions of use and storage.

Conditions to Avoid: Avoid heat, flames, sparks, and other sources of ignition. May ignite or explode

on contact with combustible materials. Keep out of water supplies and sewers.

Incompatible Materials: Hydrochloric acid is incompatible with cyanides, metals, amines, bases, metal

carbide, oxidizing materials, acids, halo carbons, combustible materials, halogens, and metal salts. Beryllium chloride is incompatible with metals.

Fire/Explosion Information: See Section 5, "Fire Fighting Measures".

Hazardous Decomposition: Thermal decomposition of hydrochloric acid may produce acid halides. Thermal

decomposition of beryllium chloride may produce acid halides and oxides of

beryllium.

Hazardous Polymerization: Will Occur X Will Not Occur

11. TOXICOLOGICAL INFORMATION

Route of Entry: X Inhalation X Skin X Ingestion

Toxicity Data

Hydrochloric Acid: Man, Oral LD_{LO}: 2 857 μg/kg

Woman, Oral LD_{LO}: 420 µl/kg

Human, Inhalation LC_{LO} : 1 300 ppm/30 min Human, Inhalation LC_{LO} : 3 000 ppm/5 min Rat, Inhalation LC_{50} : 3 124 ppm/1 h

Toxicity Data

Beryllium Chloride: Human, Inhalation TC_{LO} : 1 mg/m³

Rat, Oral LD₅₀: 86 mg/kg

Rat, Intraperitoneal LD₅₀: 5 320 μg/kg

Mutagenic, Tumorigenic,

Reproductive Data: Beryllium chloride has been investigated as a tumorigenic, mutagenic, and

reproductive effector. Hydrochloric acid has been investigated as a mutagenic

and reproductive effector.

Health Effects

(Acute and Chronic): See Section 3: "Hazards Identification" for potential health effects.

12. ECOLOGICAL INFORMATION

Ecotoxicity Data

Hydrochloric Acid: Fish: Fathead minnow (Pimephales promelas) LC₅₀ (mortality): 21 900 μg/L (96 h)

Invertebrate: Water flea (Daphnia magna) EC_{50} (immobilization): 560 µg/L (48 h) Algae: Green algae (Chlorella pyrenoidosa) EC_{50} (population size reduction):

800 μg/L (1 600 weeks)

Phytotoxicity: Water-hyacinth (Eichhornia crassipes) (residue): 1 000 µg/L (4 weeks to 48 weeks)

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Beryllium Chloride: Fish: Fathead minnow (Pimephales promelas) LC₅₀ (mortality): 150 μg/L

(96 h)

Invertebrate: Oligochaete (Lumbriculus variegatus) LC₅₀ (mortality):

 $> 10~000~\mu g/L~(96~weeks)$

Environmental Summary: Highly toxic to aquatic life.

13. DISPOSAL CONSIDERATIONS

Waste Disposal: Dispose in accordance with all applicable federal, state, and local regulations.

Hydrochloric acid is subject to disposal regulations U.S. EPA 40 CFR 262, Hazardous Waste Number D002. Beryllium chloride is subject to disposal

regulation U.S. EPA 40 CFR 262, Hazardous Waste Number D002.

14. TRANSPORTATION INFORMATION

SRM 4325: Beryllium-10/Beryllium-9 Isotopic Ratio Standard (25 mL \times 2).

U.S. DOT and IATA: Corrosive Liquid, Dangerous Goods in Excepted Quantities (25 mL × 2),

UN1789, Hazard Class 8.

15. REGULATORY INFORMATION

U.S. Regulations: CERCLA Sections 102a/103 (40 CFR 302.4):

Hydrochloric Acid: RQ 2272.73 kg (5 000 lbs) (liquid)

Beryllium Chloride: RQ 0.4545 kg (1 lb)

SARA Title III Section 302 (40 CFR 355.30):

Hydrochloric Acid: TPQ 227.27 kg (500 lbs) (gas)

SARA Title III Section 304 (40 CFR 355.40):

Hydrochloric Acid: RQ 2272.73 kg (5 000 lbs) (gas)

SARA Title III Section 313 (40 CFR 372.65):

Hydrochloric Acid: except non-aerosol forms

Beryllium and compounds

OSHA Process Safety (29 CFR 1910.119):

Hydrochloric Acid: TQ 2272.73 kg (5 000 lbs) (gas)

California Proposition 65: Beryllium and beryllium compounds are known to the state of California to cause the following: Cancer (1987).

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

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ACUTE: Yes. CHRONIC: Yes.

FIRE: No. REACTIVE: No. SUDDEN RELEASE: No.

CANADIAN Regulations: WHMIS Classification: Not determined.

EUROPEAN Regulations: EC Classification:

T Toxic

C Corrosive

Xi Irritant

N Dangerous for the Environment Carcinogen Category 2

EC Risk Phrases:

R 23/48 Toxic: danger of serious damage to health by prolonged

exposure through inhalation.

R 25 Toxic if swallowed.

R 26 Very toxic by inhalation.

R 35 Causes severe burns.

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EC Risk Phrases	continued:
D 26/27/20	T '

R 36/37/38	Irritating to eyes, respiratory system, and skin.
R 43	May cause sensitization by skin contact.
R 49	May cause cancer by inhalation.
R 51/53	Toxic to aquatic organisms, bay cause long-term adverse effects in the aquatic environment.
EC Safety Phras	ees:
S1/2	Keen locked up and out of reach of children

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EC Safety I mases.		
S1/2	Keep locked up and out of reach of children.	
S 9	Keep container in a well-ventilated place.	
S 26	In case of contact with eyes rinse immediately with plenty of water and seek medical advice.	
S 36/37/39	Wear suitable protective clothing, gloves, and eye/face protection.	
S 45	In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).	
S 53	Avoid exposure. Obtain special instruction before use.	
S 61	Avoid release to the environment.	

National Inventory Status

U.S. Inventory (TSCA): Hydrochloric acid and beryllium chloride are listed on inventory.

TSCA 12 (b)

Export Notification: Not listed.

16. OTHER INFORMATION

Sources: MDL Information Systems, Inc., MSDS *Hydrochloric Acid* 09 December 2004.

MDL Information Systems, Inc., MSDS Beryllium Chloride 16 September 2004.

Disclaimer: Physical and chemical data contained in this MSDS are provided only for use as a guide in assessing the hazardous nature of the material. The MSDS was prepared carefully, using current references; however, NIST does not certify the data in the MSDS. The certified values for this material are given in the NIST Certificate of Analysis.

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